

B	I	N	G	O
pH	Arrhenius acid	HBr(aq)	conjugate base	NO ₃ -(aq)
H ₂ SO ₃ (aq)	monoprotic acid	amphoteric	Arrhenius base	conjugate acid
equivalence point	Brønsted-Lowry acid	FREE	acid-base indicator	HNO ₃ (aq)
Lewis acid	diprotic acid	Lewis base	binary acid	end point
HNO ₂ (aq)	HF(aq)	Na ⁺ (aq)	Brønsted-Lowry base	H ₂ S(aq)

B	I	N	G	O
acid-base indicator	$\text{NO}_3\text{-(aq)}$	Brønsted-Lowry acid	equivalence point	Brønsted-Lowry base
conjugate base	$\text{HNO}_3\text{(aq)}$	Lewis acid	monoprotic acid	Arrhenius base
strong acid	$\text{Na}^+\text{(aq)}$	FREE	$\text{H}_2\text{SO}_3\text{(aq)}$	HBr(aq)
binary acid	conjugate acid	amphoteric	diprotic acid	$\text{H}_2\text{S(aq)}$
Arrhenius acid	pH	weak acid	$\text{H}_2\text{Se(aq)}$	HF(aq)

B	I	N	G	O
acid-base indicator	$\text{H}_2\text{SO}_3(\text{aq})$	binary acid	$\text{H}_2\text{Se}(\text{aq})$	Arrhenius base
Arrhenius acid	weak acid	strong acid	$\text{H}_2\text{S}(\text{aq})$	monoprotic acid
$\text{H}_2\text{SO}_4(\text{aq})$	diprotic acid	FREE	Lewis base	conjugate acid
$\text{Na}^+(\text{aq})$	Lewis acid	$\text{HNO}_3(\text{aq})$	Brønsted-Lowry acid	end point
Brønsted-Lowry base	equivalence point	$\text{NO}_3^-(\text{aq})$	$\text{HBr}(\text{aq})$	amphoteric