

B	I	N	G	O
monoprotic acid	Arrhenius acid	$\text{H}_2\text{SO}_4(\text{aq})$	conjugate base	Lewis acid
$\text{H}_2\text{S}(\text{aq})$	equivalence point	amphoteric	Arrhenius base	conjugate acid
end point	strong acid	FREE	Bronsted-Lowry acid	$\text{HNO}_2(\text{aq})$
$\text{HNO}_3(\text{aq})$	diprotic acid	Lewis acid	$\text{HF}(\text{aq})$	acid-base indicator
$\text{HBr}(\text{aq})$	Bronsted-Lowry base	binary acid	weak acid	$\text{H}_2\text{SO}_3(\text{aq})$

B	I	N	G	O
strong acid	Arrhenius base	conjugate base	H ₂ S(aq)	amphoteric
H ₂ SO ₄ (aq)	weak acid	Lewis acid	Bronsted-Lowry base	conjugate acid
HNO ₃ (aq)	Bronsted-Lowry acid	FREE	HNO ₂ (aq)	Arrhenius acid
HF(aq)	binary acid	monoprotic acid	H ₂ SO ₃ (aq)	diprotic acid
Lewis acid	acid-base indicator	end point	equivalence point	HBr(aq)

B	I	N	G	O
Bronsted-Lowry base	binary acid	Arrhenius base	acid-base indicator	Bronsted-Lowry acid
monoprotic acid	HNO ₃ (aq)	Lewis acid	equivalence point	Lewis acid
end point	H ₂ SO ₃ (aq)	FREE	diprotic acid	HF(aq)
strong acid	conjugate acid	conjugate base	weak acid	H ₂ SO ₄ (aq)
HNO ₂ (aq)	Arrhenius acid	amphoteric	H ₂ S(aq)	HBr(aq)