

B	I	N	G	O
strong acid	$-\log_{10}[\text{H}]$	$\text{C}_2\text{H}_3\text{O}_2$	pH pOH	amphoteric
7	pink	greater than 7	0-14	pH
pOH of 4.8	weak acid	FREE	1×10^{-14}	OH^-
1×10^{-9}	Polyprotic Acid	pH of 9.2	$[\text{OH}^-][\text{H}]$	pKa & pKb
Henderson Hasselbalch	completely ionize	H or H_3O^+	1×10^{-3}	less than 7

B	I	N	G	O
weak acid	pH of 9.2	pH	less than 7	pKa & pKb
1×10^{-14}	1×10^{-9}	amphoteric	7	completely ionize
0-14	Kw	FREE	$-\log_{10}[\text{H}^+]$	$[\text{OH}^-][\text{H}^+]$
strong acid	1×10^{-3}	weak base	pH pOH	greater than 7
C ₂ H ₃ O ₂	OH ⁻	Polyprotic Acid	Henderson Hasselbalch	pink

B	I	N	G	O
7	pKa & pKb	Polyprotic Acid	pH pOH	$-\log_{10}[\text{H}]$
weak acid	pH	C ₂ H ₃ O ₂	1×10^{-9}	completely ionize
OH ⁻	K _w	FREE	1×10^{-14}	0-14
1×10^{-3}	pOH of 4.8	amphoteric	H or H ₃ O	less than 7
pink	strong acid	[OH ⁻][H ⁺]	Henderson Hasselbalch	Buffer